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Avogadro's Number - the mole = the

number 6.022×10^{23} Molar Mass the

mass of one mole of any pure substance

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molar mass of an element (g/mol) equals the atomic mass of the element in grams
molar mass of a compound equals the sum of the molar masses of the elements making up the compound .
example problems: Find the molar masses of ...

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Avogadro's Number - the mole = the number 6.023×10^{23} Molar Mass the mass of one mole of any pure substance molar mass of an element (g/mol) equals the atomic mass of the element in grams molar mass of a compound equals the sum of the molar mass of the element making up the compound example problems ...

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Introduction to the Mole and Molar Mass:
The terms mole and molar mass are
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Aired: 01/07/02

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**How many formula units are in 32.6
grams of potassium oxide?**

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